

### Standard Redox Potentials of the Alkali Metals

Equation	$E_0$ value (V)
$\text{Li}^+ + \text{e}^- \rightarrow \text{Li}$	-3.05
$\text{Na}^+ + \text{e}^- \rightarrow \text{Na}$	-2.71
$\text{K}^+ + \text{e}^- \rightarrow \text{K}$	-2.94
$\text{Rb}^+ + \text{e}^- \rightarrow \text{Rb}$	-2.94
$\text{Cs}^+ + \text{e}^- \rightarrow \text{Cs}$	-3.02